

Processing thin but robust electrolytes for solid-state batteries

Are solid electrolytes stable during SSB cell operation?

In addition to being electrochemically stable during cell operation, solid electrolytes should be mechanically robust and as thin as possible to maximize the valuable volume remaining for the electrodes in SSB cell designs.

What is a high-energy-density solid-state battery (SSB)?

Nature Energy 6, 227-239 (2021) Cite this article The widespread adoption of high-energy-density solid-state batteries (SSBs) requires cost-effective processing and the integration of solid electrolytes of about the same thickness as the polymer-membrane separators found in conventional lithium-ion batteries.

Are all-solid-state lithium metal batteries suitable for high-safety batteries?

Learn more. All-solid-state lithium metal batteries (ASSLMBs) are considered as the most promising candidates for the next-generation high-safety batteries. To achieve high energy density in ASSLMBs, it is essential that the solid-state electrolytes (SSEs) are lightweight, thin, and possess superior electrochemical stability.

Can a solid electrolyte be used as a film?

It is evident from Fig. 3b that there are ample opportunities to integrate solid electrolytes even as films for a wide set of Li conductors; however, the thermal processing window of the phase and deposition selection will determine their properties and, ultimately, the integration of a given Li conductor into any commercial cell design.

Are ceramic films 'thin' electrolytes of future SSBs?

Opportunities in ceramic processing to define the chemistry and attain a deeper understanding of the relation between the structure, phase and Li-ion transport for these films will play a role in determining their integration as 'thin' electrolytes of future SSBs (Fig. 3c-e). Fig. 3: Properties of diverse oxide solid electrolytes.

Which electrolyte is used in a lithium ion battery?

All-solid-state lithium ion battery using garnet-type oxide and Li_3BO_3 solid electrolytes fabricated by screen-printing. J. Power Sources 238, 53-56 (2013). 22. Kotobuki, M. et al. A novel structure of ceramics electrolyte for future lithium battery. J. Power Sources 196, 9815-9819 (2011). 23. Kim, K. J. & Rupp, J. L. M.



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